

CHEMISTRY
WORKSHEET 3a.

Name: _____

1. Write the formulas of each of the following compounds:

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|--------------------|--------------------------|-----------------------|
| a. sodium iodide | e. potassium selenide | i. potassium fluoride |
| b. calcium sulfide | f. barium fluoride | j. potassium oxide |
| c. barium bromide | g. manganese(II) bromide | k. magnesium iodide |
| d. magnesium oxide | h. sodium chloride | l. iron(III) oxide |

2. Write the names of each of the following ionic compounds:

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|----------------------------|--------------------------|
| a. KI | g. CuCl |
| b. Co_2S_3 | h. CuCl_2 |
| c. Mg_3N_2 | i. NiBr_2 |
| d. CaBr_2 | j. SnF_2 |
| e. AlCl_3 | k. Na_2O |
| f. K_2Se | l. AuCl_3 |